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Contribution to the Calculation of Physical Properties of BeSe Semiconductor

Mohamed Amine Ghebouli and Brahim Ghebouli

Abstract

We expose various physical parameters of binary compound BeSe in the stable zinc blend and NiAs structures using the functional HSE hybrid, GGA-PBE, and LDA. We deduce elastic constants, mechanical parameters, and wave velocities according to different orientations. BeSe semiconductor has Γ -X (2.852 eV) and Γ -K (0.536 eV) bandgap in zinc blend and NiAs structures. Electrons transit from Se-p site to the Be-s state and show covalent bonding. Optical absorption peaks result from electronic transitions under ultraviolet light irradiation.

Keywords: CASTEP, zinc blend, NiAs, band structure, bandgap, absorption, BeSe

1. Introduction

Researchers have focused on II-VI semiconductors, which are intended for technological applications. BeSe crystallizes in zinc blend structure at room temperature and its properties depend on structure, pressure, temperature, and calculation method. The HSE hybrid underestimates the lattice parameters compared to GGA-PBE and LDA but gives a bandgap in accordance with the experimental value. The phonon dispersion curve of BeSe shows two optical and acoustic couplets and confirms its stability. The maximum of reflectivity (63% and 56%) is observed in the ultraviolet light domain. We expose the detailed calculation concerning the lattice parameters, elastic constants, bandgap, phonons frequencies, and optical parameters. For works carried out by other researchers, we note the transition from zinc blend phase to NiAs structure at 55 GPa [1]. The study of elastic constants and bandgap of BeSe in zinc blend phase [2, 3], the experimental bandgap ranging from 2 eV to 4.5 eV [4, 5], a small ionic radius ratio, a high degree of covalent bonding, and high hardness [6]. Appropriate bandgap and optical absorption parameters make BeSe as an important material in the area of catalysis and luminescent devices [7–9].

2. Computational method

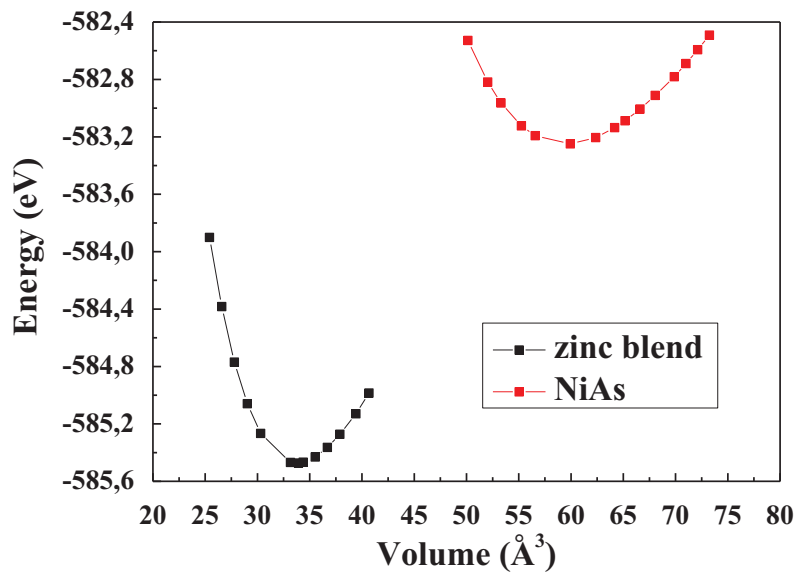
We perform calculations on BeSe by using the CASTEP code [10]. The energy E_{cut} of 660 eV and k-points of $8 \times 8 \times 8$ using Monkhorst-Pack scheme [11] ensure well

convergence. We use $20 \times 20 \times 20$ k-points in the computation of optical parameters. The GGA, LDA, and HSE hybrid functional [12, 13] treat the exchange-correlation potential. Broyden-Fletcher-Goldfarb-Shanno (BFGS) minimization technique [14] determines the structural parameters. The tolerance of geometry optimization was a total energy 5×10^{-6} eV/atom, maximum ionic Hellmann-Feynman force 0.01 eV/Å, and maximum stress 0.02 eV/Å³.

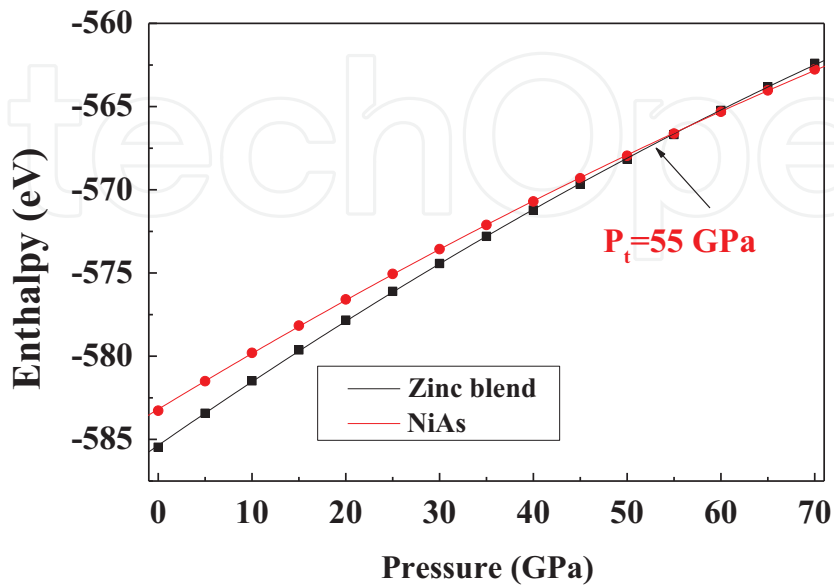
3. Results and discussions

3.1 Ground state

The position of Be atom is the corner (0, 0, 0), while the Se atom takes place at (1/4, 1/4, 1/4) and (1/3, 2/3, 1/4) for zinc blend and NiAs phases. **Figure 1(a)** gives



(a)



(b)

Figure 1. The plots of total energy versus volume (a) and enthalpy versus pressure (b) for BeSe in zinc blend and NiAs phases.

	Zinc blend					NiAs				
	GGA	LDA	HSE hybrid	Exp	Other	GGA	LDA	HSE hybrid	Exp	Other
a (Å)	5.129	5.083	4.976	5.139 [15]	5.137 [16] 5.037 [17] 5.087 [18]	3.513	3.456	3.415		3.524 [16] 3.421 [17] 3.502 [19]
c (Å)						5.459	5.461	5.347		5.462 [16] 5.398 [17] 5.529 [19]
B_0 (GPa)	74.50	87.35	99.484	92.20 [15]	74.97 [18] 80 [4] 96.5 [20]	89.20	100.30	112.76		94.7 [17] 86 [21] 109.6 [22]
B'	3.82	3.65	3.565		4.02 [23] 3.58 [24] 3.85 [19]	3.94	3.826	3.749		3.852 [17] 3.70 [25]

Table 1. Lattice parameters, bulk modulus, and its pressure derivative for BeSe in zinc blend and NiAs structures within GGA-PBE, LDA, and HSE hybrid functional.

the plots of total energy versus unit cell volume, where the zinc blend phase is more stable. **Figure 1(b)** visualizes the effect of pressure on formation enthalpy. The two enthalpy-pressure diagrams show a transition region around 55 GPa [1]. We report in **Table 1** that the lattice parameters, bulk modulus, and its pressure derivative are calculated with GGA, LDA, and HSE hybrid. Our results agree well with experimental measurements [15] and theoretical data [4, 16–25]. The HSE hybrid underestimates the lattice parameters.

3.2 Elastic constants and mechanical parameters

Table 2 includes elastic moduli of BeSe in both phases within GGA and LDA. Our elastic constants of zinc blend structure agree with theoretical values [18, 22, 26, 27]. The following relationships [28, 29] ensure the elastic stability of BeSe.

Zinc blend

$$0 < C_{11} + 2C_{12}, 0 < C_{44}, 0 < C_{11} - C_{12}, C_{12} < B < C_{11} \quad (1)$$

NiAs

$$0 < C_{11} - |C_{12}|, 0 < C_{33}(C_{11} + C_{12}) - 2C_{13}^2, 0 < C_{44} \quad (2)$$

Figure 2 shows the dependence of elastic moduli on pressure for beryllium selenide in zinc blend and NiAs phases. We list in **Table 2** the bulk modulus, shear modulus, Young's modulus, Poisson's ratio, anisotropy factor, and B_H/G_H ratio. The Poisson's ratio and anisotropy factor indicate the covalent bonding and strong anisotropy in BeSe. The reported B_H/G_H value translates the brittle zinc blend phase and ductile NiAs structure. **Figure 3** visualizes the effect of orientation on Poisson's ratio (a) and shear modulus (b) in xy, xz, and yz planes. It can be seen that the two parameters are isotropic only in the B8 phase and in the xy plane. **Table 3** predicts the maximum and minimum values of Young's modulus, linear compressibility, shear

		Zinc blend			NiAs	
		GGA	LDA	Other	GGA	LDA
C_{11} (GPa)		120.06	137.22	117 [18] 145 [24]	140.27	167.04
C_{12} (GPa)		30.73	55.94	45.5 [26] 51 [22]	71.51	80.20
C_{44} (GPa)		66.2	74.35	73.39 [26] 61 [22]	8.48	14.57
C_{13} (GPa)					36.45	46.51
C_{33} (GPa)					178.40	202.79
Θ_D (K)		460.91	465.77	407 [26]	280.75	332.74
B (GPa)	B_V	60.51	83.08		83.08	98.15
	B_R	60.51	83.08		83.08	98.15
	B_H	60.51	83.08		83.08	98.14
G (GPa)	G_V	131.16	60.86	66.24 [27]	31.23	60.86
	G_R	127.51	55.83	61.14 [27]	16.22	55.83
	G_H	129.34	58.53	63.78 [27]	23.73	58.53
E_H (GPa)		129.34	141.82	153.42 [27]	65.00	86.89
σ_H		0.138	0.21	0.203 [27]	0.36	0.35
A^U		0.188	0.45		4.62	2.60
B_H/G_H		0.46	1.41	1.349 [27]	3.5	3.05

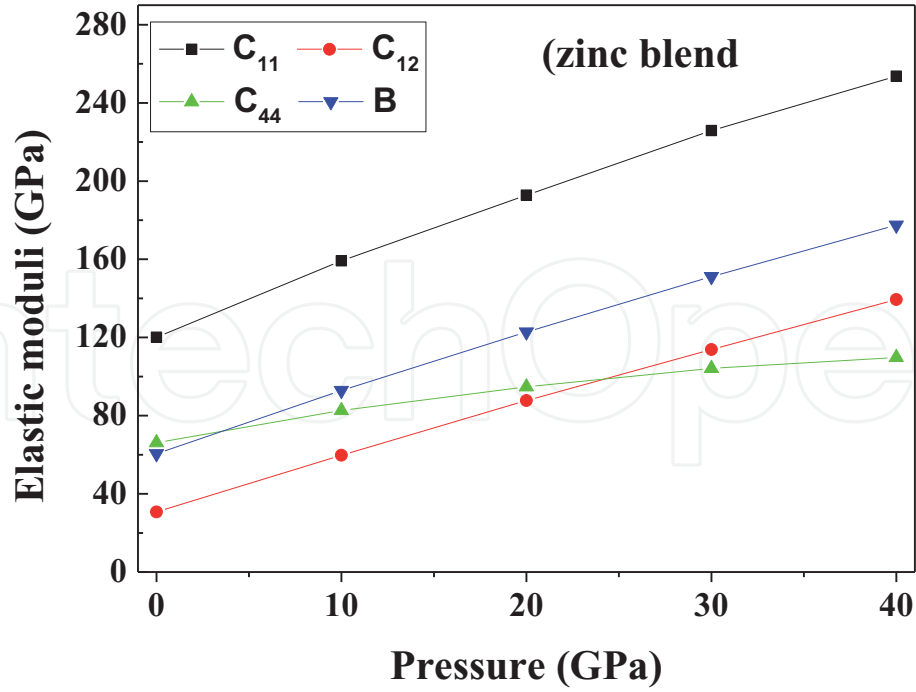
Table 2.

Elastic constants C_{ij} , Debye temperature Θ_D , bulk modulus B , shear modulus G , Young's modulus E_H , Poisson's ratio σ_H , anisotropy factor A^U and ratio B_H/G_H for BeSe in zinc blend and NiAs structures.

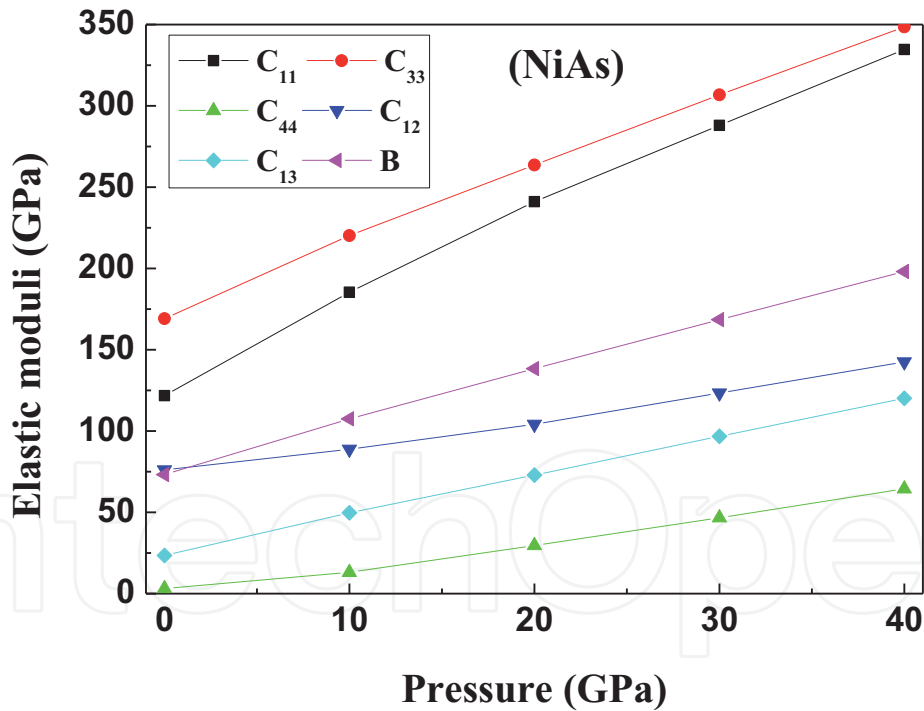
modulus, and Poisson's ratio. **Table 2** reports the Debye temperature of BeSe, which agrees with the theoretical value [26]. **Table 4** predicts elastic wave velocities along [100], [110], [111], and [001] directions.

3.3 Electronic parameters

Figure 4 shows the band structure of BeSe using GGA-PBE and HSE hybrid functional. We observe the indirect bandgap Γ -X (Γ -K) in the zinc blend (NiAs) structure. **Table 5** reports the bandgap between various symmetry points. The bandgap of BeSe in the zinc blend structure agrees with other calculations [3, 16, 23, 31]. **Figure 5** displays the dependence on the pressure of the direct and indirect bandgap. The bandgap calculated using the HSE hybrid agrees with the experimental one [3, 32, 33]. **Figure 6** shows the plots of total and partial densities of states. Se-p and Be-s orbitals are responses for the upper valence band. The first conduction band is wider for both structures and consists of Be-s and Se-p orbitals, with a small contribution of the Se-s site. The optical transition occurs from Se-p state to the Be-s site. There is a hybridization between Se-p and Be-s in the upper valence band, which translates their covalent bonding.



(a)

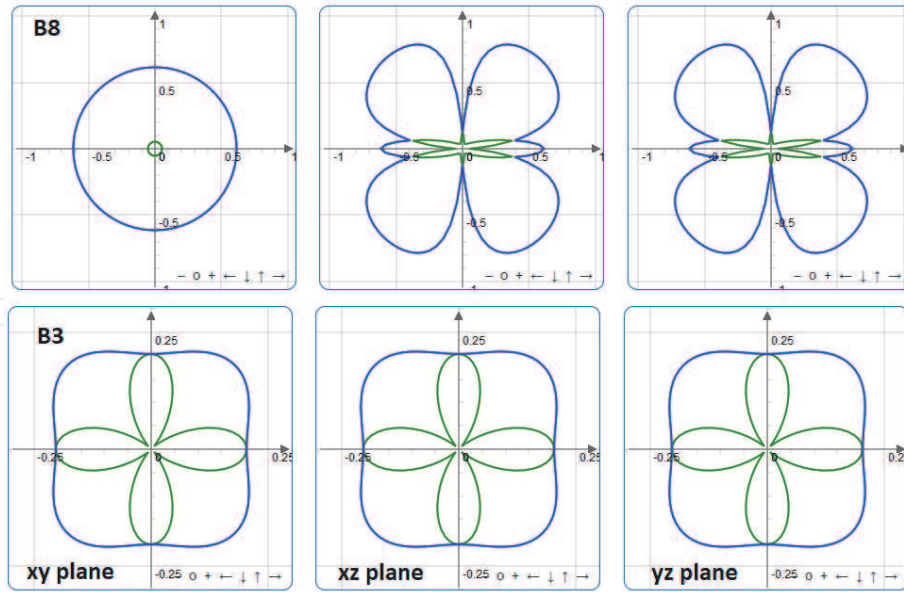


(b)

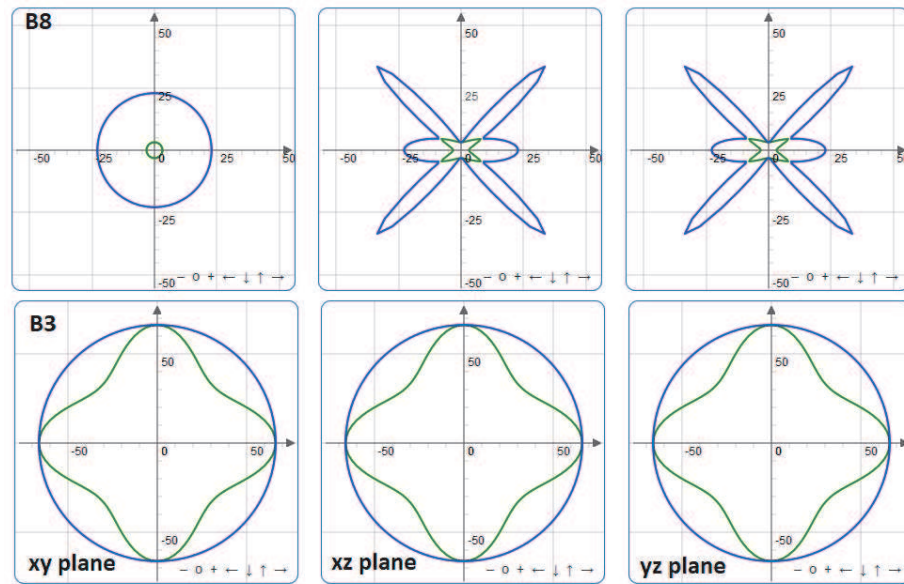
Figure 2. Elastic moduli versus pressure in zinc blend (a) and NiAs (b) phases.

3.4 Phonon frequencies

Figure 7 shows the phonon dispersion curve of BeSe in zinc blend structure. Optical (acoustic) phonons exhibit two couplets of longitudinal optical (LO)



(a)



(b)

Figure 3. The orientation effect on Poisson's ratio (a) and shear modulus (b) in zinc blend and NiAs phases.

Parameters		Young's modulus		Linear compressibility		Shear modulus		Poisson's ratio	
		E_{\min}	E_{\max}	β_{\min}	β_{\max}	G_{\min}	G_{\max}	σ_{\min}	σ_{\max}
Zinc blend	GGA	107.54	145.55	5.5083	5.5083	44.665	44.665	0.009	0.253
	LDA	104.83	171.79	4.0142	4.0142	40.644	74.354	0.0039	0.409
NiAs	GGA	42.108	163.56	4.5012	4.5012	13.201	47.33	0.118	0.623
	LDA	49.954	185.29	3.3662	3.3662	14.575	64.941	0.102	0.719

Table 3. The maximum and minimum values of Young's modulus E , linear compressibility β , shear modulus G , and Poisson's ratio σ for BeSe in zinc blend and NiAs structures.

Material	Directions	[100]	[110]	[111]	[001]
Zinc blend	v_1 (ms^{-1})	5439	5898	6043	—
	v_{t1} (ms^{-1})	3803	4305	3316	—
	v_{t2} (ms^{-1})	3803	3803	3316	—
NiAs	v_1 (ms^{-1})	5192	—	—	6067
	v_{t1} (ms^{-1})	2187	—	—	1243
	v_{t2} (ms^{-1})	1243	—	—	1243

Table 4.
 Sound velocities for BeSe along with main directions in zinc blend and NiAs structures.

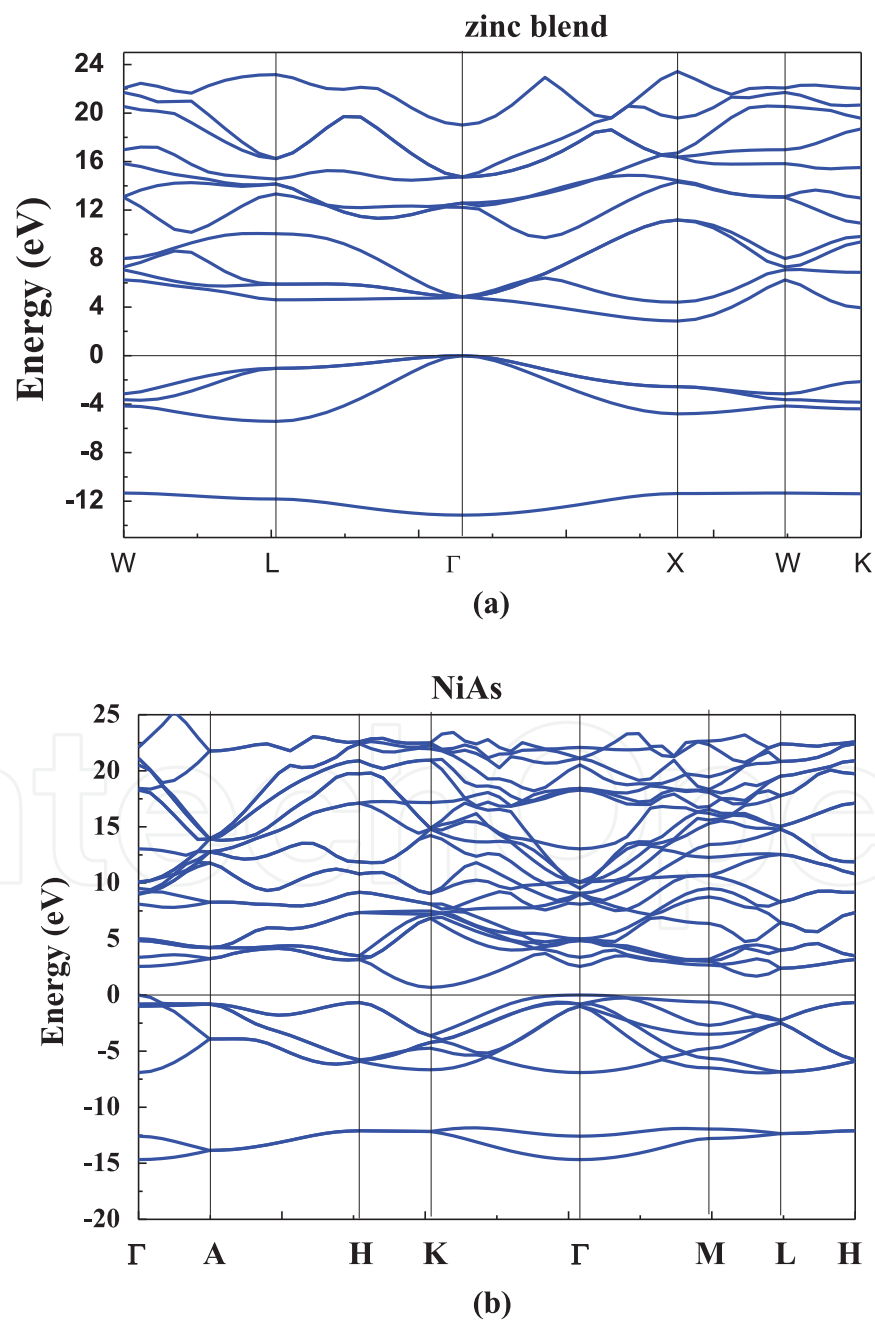


Figure 4.
 The band structure of BeSe in zinc blend (a) and NiAs (b) phases at various points in the Brillouin zone.

	Zinc blend				NiAs			
	GGA	LDA	HSE hybrid	Exp	Other	GGA	LDA	HSE hybrid
$E_{\Gamma-\Gamma}$ (eV)	4.84	4.42	5.81	4.72 [3] 5.5 [30]		2.55	2.43	3.45
E_{L-L} (eV)	5.65	5.52	7.40			4.64	4.88	6.02
E_{X-X} (eV)	5.41	5.20	6.71					
$E_{\Gamma-X}$ (eV)	2.85	2.408	3.45	4–4.5 [3]	2.41 [31] 3.12 [16]			
$E_{\Gamma-L}$ (eV)	4.60	4.39	6.05		4.33 [23]	2.38	2.27	6.23
$E_{\Gamma-K}$ (eV)	3.94	3.62	4.71	4.73 [32]		0.68	0.51	1.16
E_{M-M} (eV)						3.28	2.98	3.94

Table 5.
Band gaps for BeSe at equilibrium between various symmetry points in zinc blend and NiAs structures.

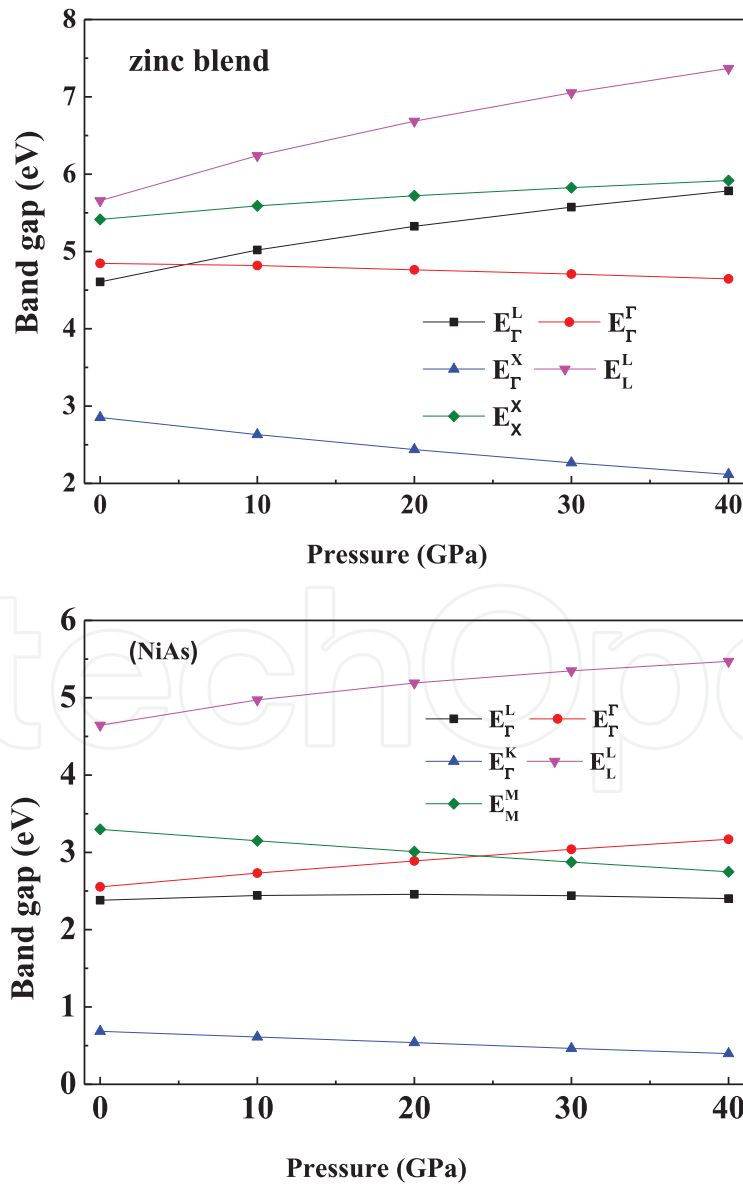


Figure 5.
The band gaps versus pressure in zinc blend and NiAs phases.

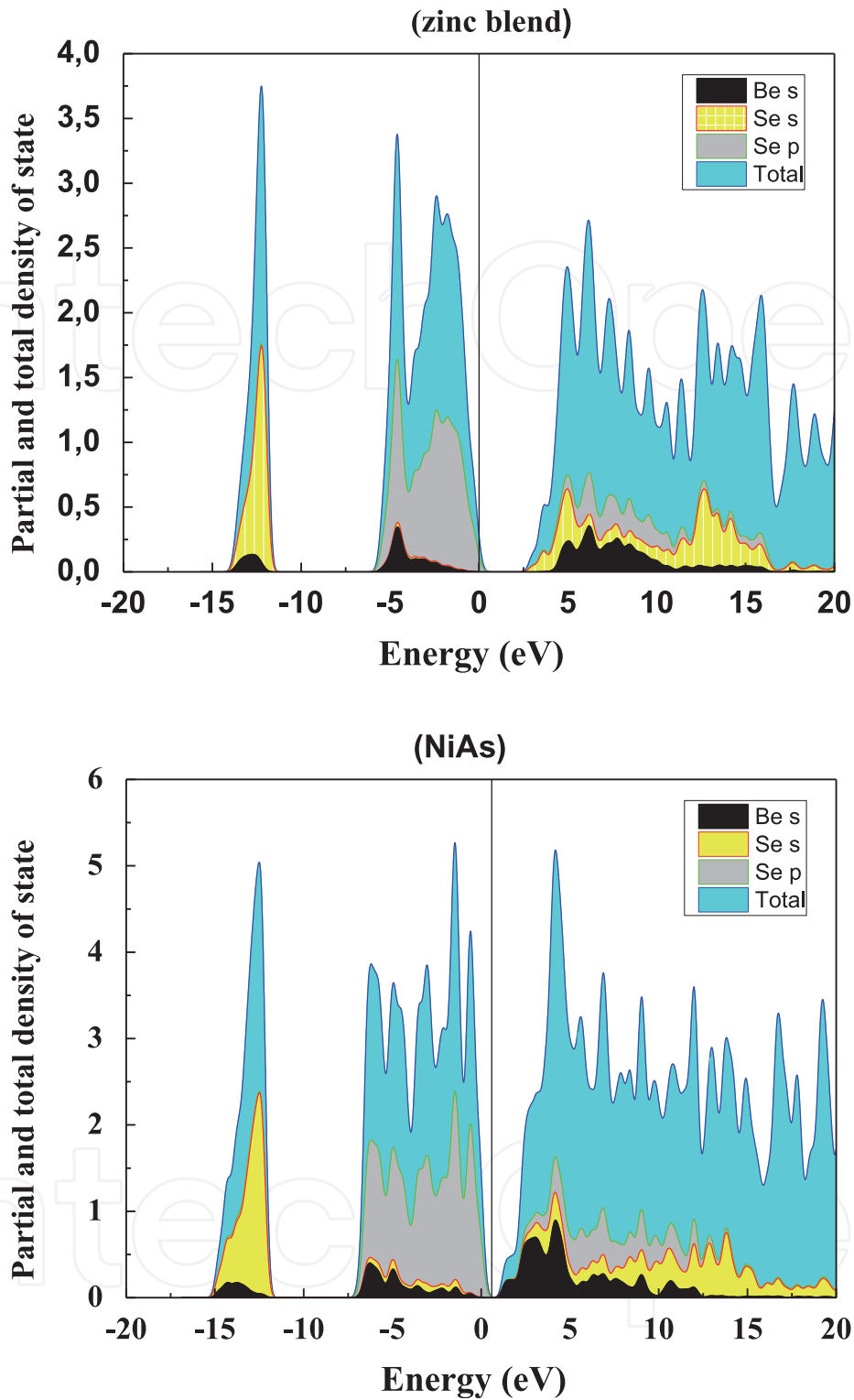


Figure 6.
 Total and partial densities of states.

{longitudinal acoustic (LA)} and transverse optical (TO) {transverse acoustic (TA)} modes and confirm the dynamical stability of BeSe. The frequency 16.06 cm^{-1} (198 cm^{-1}) separates optical (acoustic) longitudinal and transversal branches. The maximum of longitudinal optical phonons is at X point. **Figure 8** shows the effect of pressure on phonon frequencies ω_{LO} and ω_{TO} of BeSe. The equilibrium values of ω_{LO} and ω_{TO} at Γ point are 537 cm^{-1} and 471 cm^{-1} .

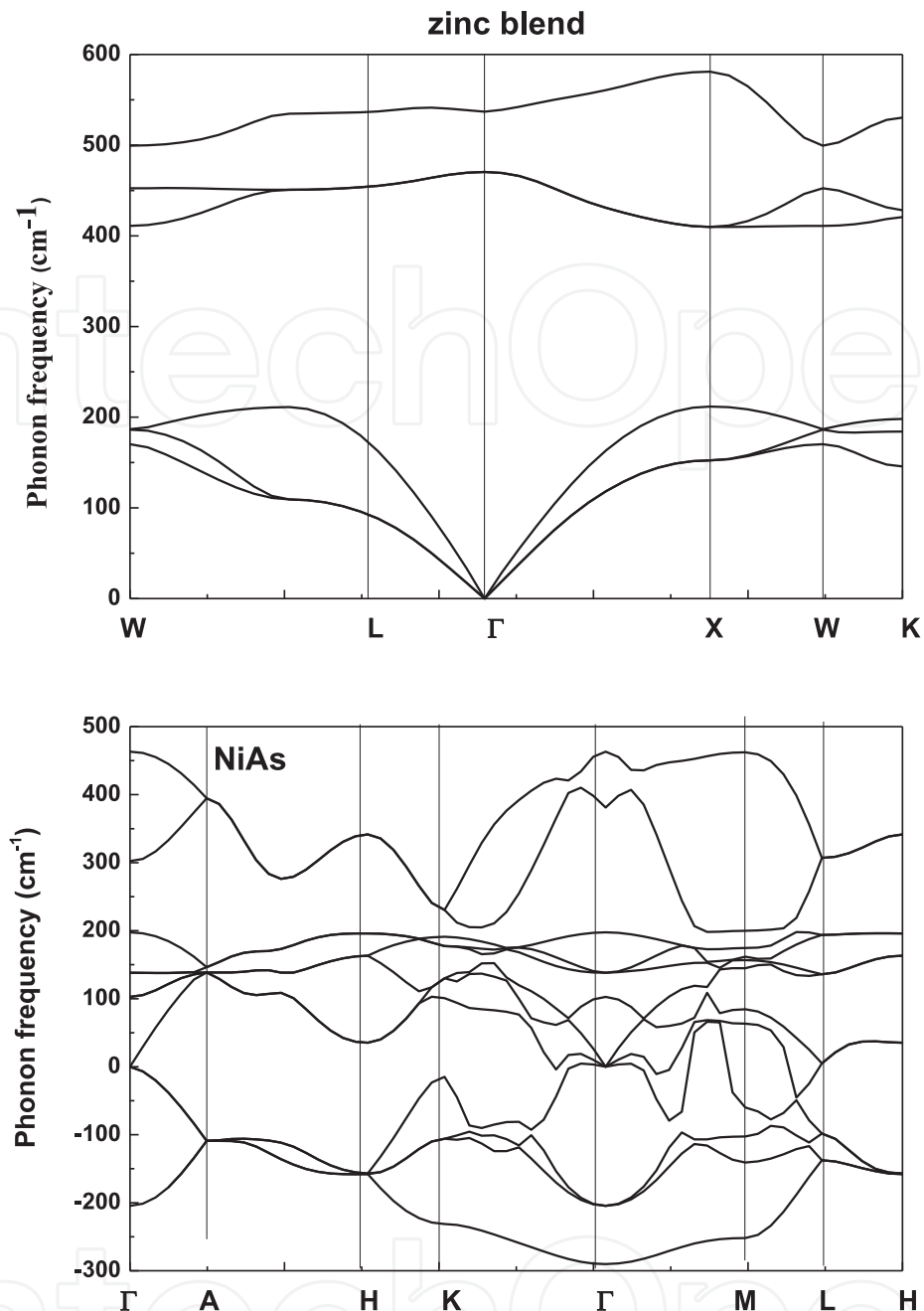


Figure 7.
The phonon dispersion curves.

3.5 Optical parameters

3.5.1 Loss function

We plot in **Figure 9** the loss function spectra versus wavelength for BeSe in zinc blend and NiAs structures. The high-loss region has a wavelength ranging from 42 nm to 90 nm. The low-loss function has a wavelength range less than 40 nm and greater than 90 nm. The maximum loss function reaches the value of 4.84% (4.49%) at 60.32 nm (57.25 nm) in the zinc blend (NiAs) structure. There is no loss in the ultraviolet and visible light domain.

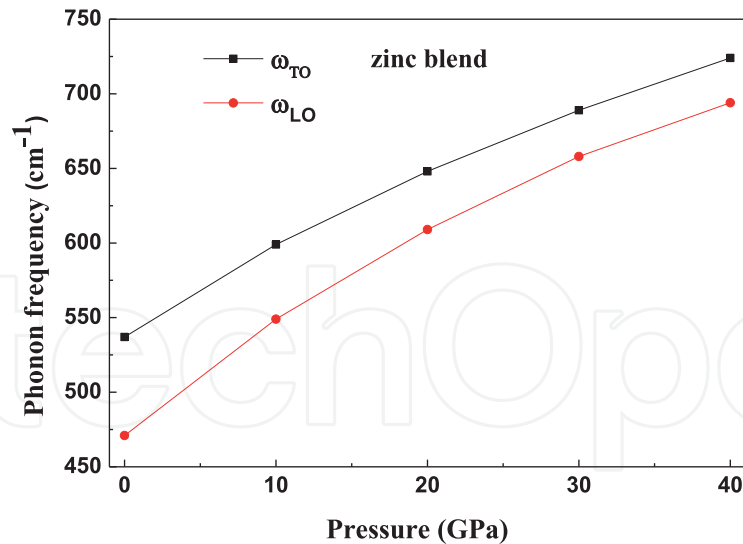


Figure 8.
The phonon frequency versus pressure at Γ point.

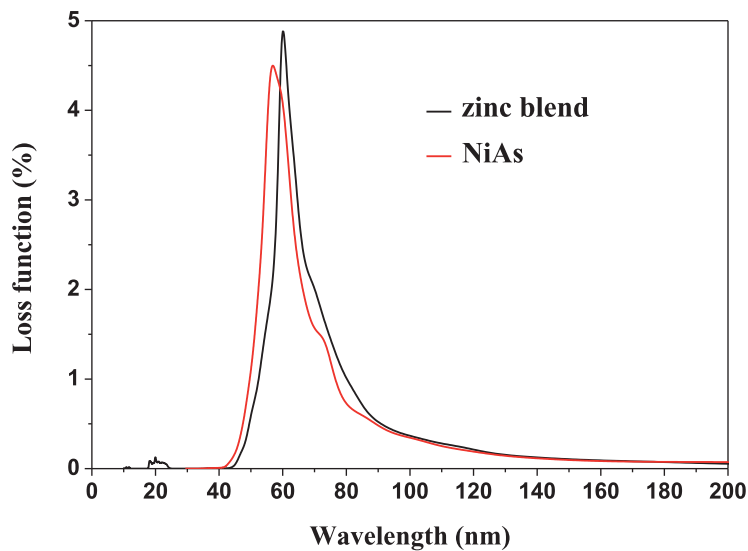


Figure 9.
The loss function versus wavelength.

3.5.2 Optical absorption

We applied a smearing value of 0.1 to obtain more distinguishable absorbance peaks. We attribute the absorption peaks as depicted in **Figure 10** to the photo transition energies from the maximum valence band to the minimum conduction band under ultraviolet light irradiation. BeSe absorbs a maximum of ultraviolet light $331,971 \text{ cm}^{-1}$ ($335,643 \text{ cm}^{-1}$) at wavelength 163.6 nm (163 nm) in zinc blend (NiAs) structure, shows narrow indirect bandgap, then it is a good candidate for photocatalysis in the ultraviolet light domain (120–400 nm).

3.5.3 Optical reflectivity

The reflectivity estimates the amount of incident light on the surface of photocatalytic material. The reflectivity of BeSe, as shown in **Figure 11**, starts at 0.007

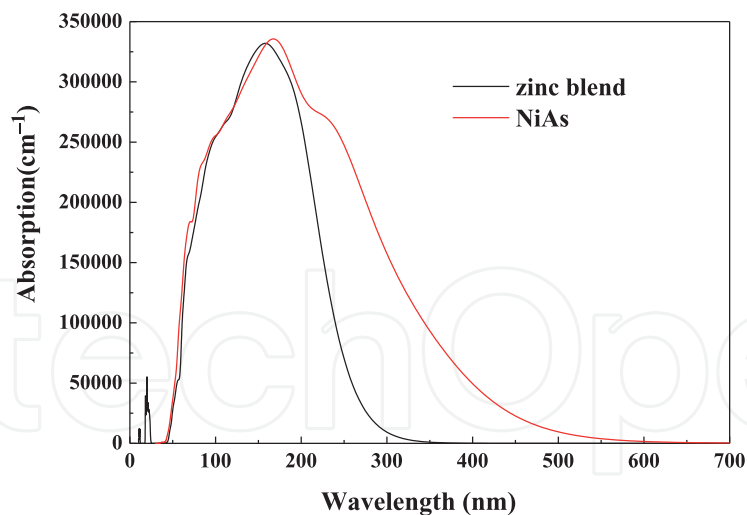


Figure 10.
The absorption versus wavelength.

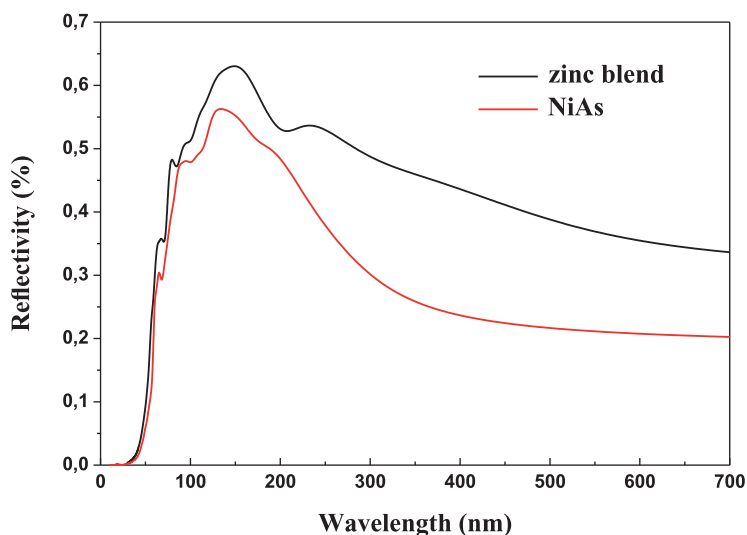


Figure 11.
The reflectivity versus wavelength.

and corresponds to a wavelength around 33 nm. It reaches a maximum value 0.63% (0.56%) at 149 nm (136.5 nm) in zinc blend (NiAs) structure. The reflectivity in the ultraviolet light domain (120 nm–400 nm) is 0.63% (0.56%) in zinc blend (NiAs) structure. While in the visible light (400 nm–800 nm), it is 0.43% (0.23%) for zinc blend (NiAs) phase.

4. Conclusion

We study BeSe in zinc blend and NiAs phases using GGA-PBE, LDA, and HSE hybrid. BeSe is a good candidate for photocatalysis material. We estimate the lattice parameters, shear modulus, Young's modulus, Poisson's ratio, average sound velocities, and Debye temperature. The maximum absorption at ultraviolet light is $331,971 \text{ cm}^{-1}$ ($335,643 \text{ cm}^{-1}$) in zinc blend (NiAs) structure. We perform transverse and longitudinal optical phonon frequencies ω_{TO} and ω_{LO} at Γ point. A frequency gap

of 16.06 cm^{-1} (198 cm^{-1}) in zinc blend structure separates longitudinal and transversal optical (acoustic) branches. The reflectivity in the ultraviolet light domain is 0.63% (0.56%) in zinc blend (NiAs) structure. While in the visible light, the reflectivity is 0.43% (0.23%) for the zinc blend (NiAs) phase.

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