



## 1st International Conference on Physico-chemistry, Materials Sciences & Applied Mathematics



# Theoretical analysis of structural, optoelectronic and magnetic properties of ABO<sub>2</sub> Delafossite semiconductor

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**KEYWORDS:** Delafossite, TB-mBJ, Magnetic properties, Dielectric function, Absorption coefficient and the reflectivity.

### **ABSTRACT**

The Delafossite group of oxide materials was initially uncovered in 1965, comprising a category of ternary compounds characterized by the general formula ABO<sub>2</sub>, with A and B denoting metallic elements. One of the most prevalent Delafossites is CuAlO<sub>2</sub>, which stands as a p-type semiconductor known for its remarkable electrical conductivity and transparency across the visible and near-infrared regions of the electromagnetic spectrum [1–3]. These distinctive qualities have propelled Delafossite materials into the spotlight for a wide array of applications, including use in electronic devices, energy storage, energy conversion, and optoelectronics. Their intriguing electronic, magnetic, and optical attributes have only recently garnered attention [4,5], prompting researchers to delve into these materials, not only to grasp their fundamental properties but also to uncover potential applications. Density functional theory (DFT) calculations have unveiled a range of electronic band structures within Delafossites, varying from metals to semiconductors contingent upon the specific choice of A and B ions. Furthermore, DFT calculations have forecasted that particular Delafossites, such as CuCrO<sub>2</sub> and CuFeO<sub>2</sub>[6,7], manifest antiferromagnetic ordering due to the interplay of spins on the transition metal ions. In an ab initio study conducted by Azmat et al. [8], it was demonstrated that CuXO<sub>2</sub> compounds (where X represents Al, Ga, In, B, La, Sc, Y) display semiconductive behavior. Experimental investigations into Delafossites have encompassed the electronic and optical properties. Among the most renowned Delafossites is CuAlO<sub>2</sub>, renowned for its high electrical conductivity and optical transparency. CuAlO<sub>2</sub> boasts exceptional carrier mobility and low carrier concentration, making it an ideal candidate for transparent conductive electrodes in optoelectronic devices. Additionally, Delafossites have undergone scrutiny for their photovoltaic potential, with CuInO<sub>2</sub> and CuGaO<sub>2</sub> exhibiting promising results for solar cell applications. An experimental study by Kushwaha et al. [9] explored the PdRhO<sub>2</sub> compound, revealing its crystallization in a rhombohedral-type structure characterized by crystal cell dimensions of a = 3.0240 Å and c = 18.096 Å.

In this study, we perform first-principles calculations based on density functional theory and the semi-classical Boltzmann method to investigate the structural, optoelectronic, magnetic, thermodynamic and thermoelectric properties for BrCdO<sub>2</sub> in the Tetragonal phase. To compute the structural properties, the Wang and the Perdew-Burke-Ernzerhof generalized gradient approximation (PBE-GGA) was used as exchange-correlation potentials. Besides, the modified Becke-Johnson functional of Tran and Blaha (TB-mBJ) are used to compute the electronic and optical properties to achieve the best band gap energy values and a higher degree of precision. Our calculations have revealed that this compound have direct band gap 4.17 eV. As a result of this study, BrCdO<sub>2</sub> is promising material for optoelectronic devices, especially as photovoltaic materials in solar cells.

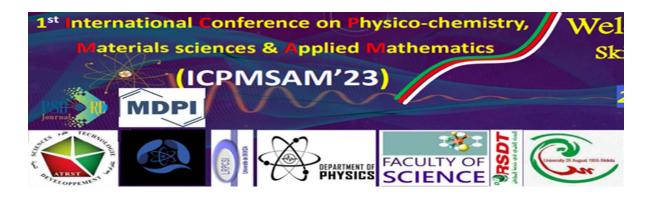


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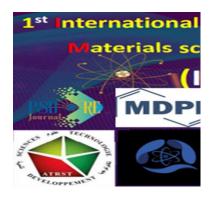


	11/27/2023							
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09:00-09:05	Opening	5						
09:05-10:005	Protoco	l						
	Plenary	session ( chairman Louidi)						
10:15-10:45								
10:45-11:15	Pr.MERA	AD Ahcen						
11:15-11:44	Pr.Pierre	e francois brevet (online) lyon Fra	ince					
11:45-12:25	Pr.CONS	STANTINESCU Catalin-Danie (onlir	ne) Marseille	France				
12:00-13:00	Coffee b	oreak	Coffee break	Coffe	ee break			
		Chairman			Chairman			
Time	Code	Room1	Time	Code	Room2			
13:00-13:20		MOSBAH Salima	13:00-13:20		HADEF Zakaria			
13:20-13:40		BOUSBA Houssem Eddine	13:20-13:40		BOUDIAR Abid			
13:40-14:00		MOHAMMED BELHADJ Ahlem H	13:40-14:00		KHENNICHE Ghania			
14:00-14:20		DJEGHLOUL Fatima	14:00-14:20		BENAISSA Amina			
14:20-14:40		LASMI Sofiane			BENTRIDI Salah eddine			
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		Chairman			Chairman			
Time		Online1	Time	Code	Online2			
17:00-17:20		ABDELDJEBAR Hasnia	17:00-17:20		DAHO Salah Eddine			
17:20-17:40		CHIKHI Sara	17:20-17:40		SAADIYA Benatmane			
17:40-18:00		ZIANE Mohamed Issam	17:40-18:00		KHARCHI Nadia			
	11/28/2023							
09:00-09:30		Pr.CHIHA Mahdi						
09:30-10:00	Pr.Haddad Salim							
10:00-10:30		Pr.Bouzred Hamoudi						
		Pr.Maazouzi Ismaine						
Time		room1	Time	Code	room2			
10:30-11:00		BOUDJEDAA Tahar			BENAISSA Amina			
11:00-11:20		BENHACHEM F,Zahra	11:00-11:20		HOUDA Amina			

11:20-11:40		OTMANI Sadok	11:20-11:40		AOUICI Samia
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13:00-13:20		BENKHANOUCHE Zoubir	13:00-13:20		KEHAL Youcef
13:20-13:40		MESSAI Ridha	13:20-13:40		KEHAL Abir
13:40-14:00		KEBAILI Selwa	13:40-14:00		MOURES Nadjiba
14:00-14:20		HARIDI Ryad	14:00-14:20		CHERIET Loubna
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	Chairman			Chairman
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17:00-17:20	BENAKCHA Mansoura	17:00-17:20		SAADIYA Benatmane
17:20-17:40	LAKRI Amel	17:20-17:40		Djilali Khadidja
17:40-18:00	HANNACHI Ibtissem	17:40-18:00		





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13:00-13:20		LAOUET Nadjet	13:00-13:20		ZAIEM Slimane
13:20-13:40		BEGAG Abdelaziz	13:20-13:40		BENHACHEM Fatim
13:40-14:00		LAOUER Abdelghani	13:40-14:00		SID-SAHTOUT Nazil
14:00-14:20		BOUADI Abdelkader	14:00-14:20		MENASRI Abdellah
14:20-14:40		REZIG Walid	14:20-14:40		
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		Chairman			
Time	Code	Online 3			
17:00-17:20		SLATNIA Randa			
17:20-17:40		ABDELDJEBAR Hasnia	l		
17:40-18:00		MEDJANE Chahinez			
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10:30-11:00		MERAD Mahmoud	10:30-11:00		Grimed HOURIA
11:00-11:20		ROULA Abdelmalek	11:00-11:20		Ketfi M. El Amine

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	13:00-13:20		SENOUCI Yasmine	13:00-13:20		BENABDELLAZIZ Ou
	13:20-13:40		BOUSLAH Zineb	13:20-13:40		HANNACHI Ibtissem
	13:40-14:00		BOUGHEZALA Nasrin	13:40-14:00		MOKRANI Khawla
	14:00-14:20		BENKHANOUCHE.Z	14:00-14:20		CHEBLI Fatiha
	14:20-14:40		MALAOUI Yousra	14:20-14:40		KERBICHE Hind
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13:00-13:20		BABOURI Kaoutar	13:00-13:20					
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